

## Isotopes Practice Set Answers

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**isotopes - practice worksheet** - isotopes - practice worksheet backgrou nd information : atoms of a given element which have the same number of protons but different numbers of neutrons are called isotopes. thus, isotopes have the same position in the periodic table, the same chemical properties and the same atomic charge. the simplest example of an atom with different isotopes is hydrogen. the three isotopes of hydrogen ...

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**isotope practice worksheet - chemistry** - 6. calculate the atomic mass of copper if copper-63 is 69.17% abundant and copper-65 is 30.83% abundant. answer: 63.62 amu 7. boron exists in two isotopes, boron-10 and boron-11.

**reminders: ion practice set - thomas county schools** - isotopes practice set 1. what is an isotope? 2. what does the number next to isotopes signify? 3. how can you tell isotopes apart? for each of the following isotopes, write the number of protons, neutrons, and electrons. chromium-58 chromium-63 # of protons # of neutrons # of electrons ...

**isotope and ions practice worksheet name:** - isotope and ions practice worksheet name: \_\_\_\_\_ part i: isotopes 1. define an isotope. 2. what would happen if the number of protons were to change in an atom? 3. another way to write isotopes is to write the name of the element then add the mass number after a dash, for example,  $^{14}_6\text{C}$  is carbon-14. why isn't the atomic number needed for this notation? ...

**answers to practice test questions 2 atoms, isotopes and ...** - answers to practice test questions 2 . atoms, isotopes and nuclear chemistry . 1. fluorine has only one stable isotope. its mass number is   19  .

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**isotope practice worksheet - harrischemistry** - naturally occurring europium (eu) consists of two isotopes was a mass of 151 and 153. europium-151 has an abundance of 48.03% and europium-153 has an abundance of 51.97%.

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