

## Chapter 6 Modern Chemistry Answers

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**6 chemical bonding - effingham county schools / overview** - chapter 6 review chemical bonding section 2 short answer answer the following questions in the space provided. 1. use the concept of potential energy to describe how a covalent bond forms between two atoms. as the atoms involved in the formation of a covalent bond approach each other, the electron-proton attraction is stronger than the electron-electron and proton-proton repulsions. the atoms ...

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**assessment chapter test a - kettering city school district** - modern chemistry 47 chapter test name class date chapter test a, continued \_\_\_\_\_ 7. the melting points of ionic compounds are higher than the melting points of molecular compounds because a. ionic substances tend to vaporize at room temperature. b. ionic substances are brittle. c. attractive forces between ions are greater than the attractive forces between molecules. d. the numbers of ...

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